



HOLIDAYS HOME WORK

June-July 2015-16

Class X

ENGLISH:

- English main course book chapter 1, 2
- English workbook chapter 1, 2
- Draw the web chart of the drama given in the English literature book
- Do the questions of novel which are given in the class (related to the chapters done) in the separate notebook.
- Do 1-5 listening exercises from ASL portion of dev jyoti bravia (content available in cds of bravia)

MATHS:

1. Do Activity: 7, 8, 11, 12 from Lab Manual
2. Assignment from Mamta Batra of Ch 1, 2, 3, 8, 14

SOCIAL SCIENCE:

1. Revise the syllabus done till date thoroughly
2. Prepare a report on rescue team including the different bandages (with pics) and do's and don'ts during disasters.
3. From the mentioned topics make any two projects (to be evaluated in fa's as per cce)
 - (a) Nepal's earthquake case study
 - (b) Safer construction practice
 - (c) Survival skills
 - (d) Sharing responsibilities
 - (e) Alternative communication system during disasters
4. After revising the syllabus complete the map workbook (geography & history) pages relevant to the syllabus to recapitulate.



SCIENCE:

PHYSICS

1. Completion of practical file.
2. Extra questions at least 25 from both the chapters.
 - a) Electricity
 - b) Magnetic effects of electric current.
3. Revision of all the covered syllabus.



BIO

1. Learn the Question answers of Ch-6 (Life-Processes).
2. Practice all the diagrams of Ch-6 in fair note book.
3. Frame 15 extra questions and 15 M.C.Q's from Ch-6 and also write their answers in fair note book.

Chemistry

Give reasons for the following :

- (i) Keeping food in air tight containers help in preventing rancidity.
- (ii) Moist air and acidic gases are not good for some metals.
- (iii) Manufactured chips usually flush bags of chips with nitrogen gas.
- (iv) Sodium acts as an oxidising agent while chlorine acts as a reducing agent in the following reaction :
$$2\text{Na(s)} + \text{Cl}_2\text{(g)} \longrightarrow 2\text{NaCl(s)}$$
- (v) White coloured silver chloride turns gray when kept in sunlight.

q.2 When a green iron salt is heated strongly its colour finally changes to black and odour of burning sulphur is given out.

- (i) Name the iron salt.
- (ii) Name the type of reaction that take place.
- (iii) Name the compound which changes into Brownish black colour.
- (iv) Name the compound which is responsible for its odour.
- (v) Write the chemical equation of the above statement.

q.3 A chemical is heated in a test tube brown fumes comes out and a black residue is left behind.

- (a) Name the chemical which gives brown fumes.
- (b) Write the equation.
- (c) Name the compound which gives black residue.

q.4. Give one example of decompositon reaction which is carried in the presence of :

- (i) electrical energy
- (ii) Sun light
- (iii) Heat energy

Q.5. Give two examples of everyday life situations where redox reactions are taking place.

Q.6. Why does the blue colour of copper sulphate solution change when a piece of iron is dropped into it?

Q.7. State any two ways to prevent the rancidity of food containing fats and oil?

Q.8. Why is respiration an exothermic reaction?

Q.9. **Very short answers (1 mark)**

1. What type of reaction is represented by the digestion of food in our body?
 2. When ammonium hydroxide solution is added to aluminium chloride, a white precipitate of aluminium hydroxide is formed along with ammonium chloride solution. What type of chemical reaction is stated in the above chemical change?
 3. What is a thermite reaction? Mention its use.
 4. In the refining of silver, the recovery of silver from silver nitrate solution involves a displacement reaction with copper metal. Write down the chemical equation for the reaction.
 5. What do you observe when a solution of lead nitrate and potassium iodide in water are mixed together?
 6. A white substance 'X' on exposure to sunlight turns grey. Name the substance 'X' and the grey product. \rightarrow
 7. What would you observe when lead nitrate is heated?
8. When hydrogen gas is passed over a black substance 'X', the black coating turns brown due to (i) oxidation (ii) reduction.
9. Define Rancidity.
10. What are oxidising agents? Page -03

Complete all the practical's of SA I In your practical file

Revise the chapters Chemical equations and Acid, Base & Salts.

PUNJABI:

ਚਮਾਤ -ਦਸਵੀਂ

1. ਪੰਜਾਬੀ ਸਹਿਤ ਦੇ ਪੰਜ ਸਹਿਤਕਾਰਾਂ ਸੰਬੰਧੀ ਫਾਇਲ ਤਿਆਰ ਕਰੋ।

2. ਕਾਪੀ ਤੇ ਕਰਵਾਇਆ ਗਿਆ ਸਾਰਾ ਪਾਠਕ੍ਰਮ ਯਾਦ ਕਰੋ।

HINDI:

अनुदेश:-

1. स्मार्ट कक्षा को अपडेट करना
2. रोलैट्रॉनिक मीडिया की भूमिका

पत्र:- औपचारिक

1. आपके आस-पास को दुकानों में मिलानदी सामान बेचा जा रहा है। इसकी सूचना देते हुए ब्लॉक मुख्यालय, उप पीठकपुरा, दिल्ली को और से स्वास्थ्य अधिकारी को पत्र लिखिए।
2. अपनी स्थल प्रकषीत बरताने के लिए पित्त प्रालेखन-कक्षाएक पत्र के संसाधन को पत्र लिखिए।

सूचना लेखन:-

1. अंततः अस्पताल के रक्त जांती सन्तारे पर अस्पताल अधिकारी, अधिकारा होने के नारे मुफ्त उच्च शिक्षित की सूचना देते हुए सूचना-पत्र तैयार करना।
2. आज दादी बच्चा के विद्यार्थी हैं। आपको स्कूल के बंधकट बॉल मैदान में एक दादी मिली है। इसकी सूचना देने हेतु छात्रों के लिए सूचना-पत्र तैयार कीजिए।

विज्ञापन लेखन:-

1. अपनी बहन के लिए उचित तर हूँने हेतु अवकाश में देने के लिए 20-25 शब्दों में विज्ञापन तैयार कीजिए।
2. आपको स्कूल में श्रुतिवाचकशा में एक बेंच का आयोजन मिला जा रहा है जिसमें फिल्म प्रकार की श्रुतिवाचकशा काजर्न जायगी। इस बेंच आयोजन हेतु विज्ञापन तैयार कीजिए (उचित शब्द)



COMPUTER:

1. Create a Website in www.wordpress.com on the Topic "Environment Pollution" or International Projects ,create minimum 4 Menus/links in it and mail the website link to your respective teachers

Teacher Name	E-Mail ID
Ms Kirandeep Kaur	Kirandeep_18march@yahoo.co.in
Ms Harmanpreet Kaur	harmans_110@yahoo.com
Mrs Poonam Dogra	issh786s@gmail.com
Ms Rashi Sharma	Rashi.sharma031@gmail.com

2. Learn and complete Notebook of L-6